

Name _____

Ions and Formulas For AP Chemistry

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|------------------------|-------------------------|--------------------------|
| 1. Chromic | 36. Neon | 71. Ferric |
| 2. Flouride | 37. Water | 72. Carbonate |
| 3. Sodium | 38. Chromium (III) | 73. Potassium |
| 4. Mercury (II) | 39. Barium | 74. Carbon tetrachloride |
| 5. Calcium | 40. Nitric Acid | 75. Mercury (I) |
| 6. Chloride | 41. Chlorite | 76. Hypochlorite |
| 7. Ammonia | 42. Carbon Monoxide | 77. Lead (IV) |
| 8. Iodide | 43. Magnesium | 78. Sulfate |
| 9. Sulfide | 44. Chromate | 79. Permanganate |
| 10. Hydrogen Sulfate | 45. Dichromate | 80. Nickelic |
| 11. Bisulfate | 46. Aluminum | 81. Lead (II) |
| 12. Chlorate | 47. Iron (III) | 82. Proton |
| 13. Copper (II) | 48. Hydrogen ion | 83. Sucrose |
| 14. Hydrogen | 49. Methane | 84. Chromium (II) |
| 15. Peroxide | 50. Cuprous | 85. Sulfuric Acid |
| 16. Hydroxide | 51. Ammonium | 86. Hydorchloric Acid |
| 17. Hydrogen Carbonate | 52. Stannous | 87. Flourine |
| 18. Bicarbonate | 53. Nitrite | 88. Chlorine |
| 19. Ferrous | 54. Hydronium | 89. Iodine |
| 20. Oxygen | 55. Nitrogen | 90. Lithium |
| 21. Tin (IV) | 56. Bromine | 91. Bromate |
| 22. Perchlorate | 57. Nickelous | 92. Cobalt (II) |
| 23. Beryllium | 58. Mercurous | 93. Nitride |
| 24. Zinc | 59. Plumbic | 94. Manganese (II) |
| 25. Copper (I) | 60. Acetate | 95. Tin (II) |
| 26. Iron (II) | 61. Bromide | 96. Cyanate |
| 27. Plumbous | 62. Diarsenic pentoxide | 97. Thiocyanate |
| 28. Stannic | 63. Oxide | 98. Thiosulfate |
| 29. Sodium Chloride | 64. Helium | 99. Oxalate |
| 30. Nickel (I) | 65. Glucose | 100. Cyanide |
| 31. Nickel (II) | 66. Cupric | |
| 32. Chromous | 67. Nitrate | |
| 33. Phosphate | 68. Silver | |
| 34. Mercuric | 69. Sulfite | |
| 35. Sodium Hydroxide | 70. Carbon Dioxide | |